

Enhanced Electromagnetic Wave Absorption of Three-Dimensional Porous Fe₃O₄/C Composite Flowers

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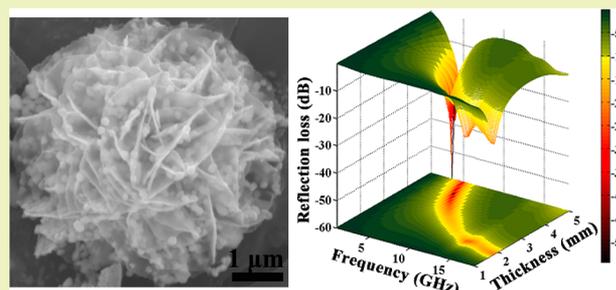
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ABSTRACT: Magnetite (Fe₃O₄)/carbon (C) composite flowers with an average size of 4–6 μm were prepared through a facile route including a solvothermal approach and a carbon reduction process. The resultant Fe₃O₄/C composites are porous and exhibit a three-dimensional (3D) flower-like morphology with the core–shell Fe₃O₄@C nanoparticles hybridized by amorphous carbon sheets. The epoxy resin composites containing 50 wt % 3D porous Fe₃O₄/C composite flowers display an optimal reflection loss (RL) value of –54.6 dB at 5.7 GHz at a thin thickness of 4.27 mm and the effective bandwidth with RL < –10 dB reaches 6.0 GHz at a thickness of 2.1 mm.

These enhanced EM wave absorption performances are attributed to the synergistic effects of Fe₃O₄ and carbon as well as the structural advantages, e.g., three-dimensional structure with large surface area, porous and core–shell structures of Fe₃O₄/C flowers. These results suggest the 3D porous Fe₃O₄/C composite flowers designed here can serve as ideal candidates for high-performance EM wave absorption.

KEYWORDS: Electromagnetic wave absorption, Composite, Magnetite, Carbon, Three-dimensional



INTRODUCTION

Nowadays, tremendous efforts have been devoted to the fabrication of electromagnetic (EM) wave absorption materials in order to solve the increasingly electromagnetic radiation pollution arising from the rapid development of electronic devices.^{1–6} As an important kind of magnetic material, Fe₃O₄ has drawn a great deal of attentions because of its low cost and unique magnetic features, such as a proper saturation magnetization value and high Curie temperature, thus providing great potential to be applied as high-efficiency EM wave absorbers.^{7–11} However, pure Fe₃O₄ absorbers always suffer from a lot of drawbacks such as a dramatic decrease of permeability in high frequency range due to the Snoek's limit, ease of oxidation, and high density, which is unable to satisfy the requirements of an ideal EM absorber with strong absorption ability, broad absorption bandwidth, thin absorber thickness and low density. One effective way to overcome these problems is to design core–shell structured composites with Fe₃O₄ as the core and a dielectric shell. Among a variety of dielectric materials, carbon materials especially at the nanoscale, for their lightweight, abundant resources, large aspect ratio and good conductivity, are always the most attractive candidates to improve EM wave absorption perform-

ances.^{12–15} These core–shell Fe₃O₄/C composites were found not only to significantly improve the impedance matching due to the synergy effect of strong magnetic loss from magnetite cores and dielectric loss from carbon shells but also to effectively reduce the density of the absorbers and protect the magnetite core from environmental oxidation, giving rise to enhanced EM wave absorption properties than their individual components. For example, Du et al. demonstrated a successful preparation of core–shell Fe₃O₄/C microspheres with different shell thicknesses and the microwave absorption properties were investigated.¹⁵ The reflection loss (RL) lower than –10 dB (90% absorption) was achieved from 4.0 to 18 GHz with a thickness range of 1.5–5.0 mm. Chen et al. have synthesized porous Fe₃O₄/C core–shell nanorods via a three-step process and the minimum RL reached –27.9 dB at 14.96 GHz attributing to their improved impedance matching.¹⁶

Apart from constructing core–shell structured composites, fabricating materials with flake structure has been another effective way to obtain high-efficiency EM absorbers. Large

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shape anisotropy of flake-shaped structure can exceed the traditional Snoek's limit, leading to a larger permeability in GHz range.^{17–19} Recently, the 3D flower-like structure assembled by abundant flakes has been promising for EM wave absorption, such as flower-like FeNi@C nanocomposites with an optimal RL of -47.6 dB at 3.17 GHz.²⁰ Lv et al. have prepared porous 3D flower-like Co/CoO with an optimal RL value of -50 dB at 7.2 GHz, and the bandwidth was up to 4.2 GHz at a thickness of 2.0 mm.¹⁸ The excellent EM absorption is associated with a better impedance matching between permeability and permittivity of the composites. Moreover, special 3D flower-like structure of Co/CoO composites with the presence of pores can cause multiple scattering and reflection, which contribute to the dissipation of incident EM waves as well. However, there are few studies concerning about the synthesis and applications of 3D flower-like structured Fe₃O₄ as EM absorbers. Li et al.²¹ have synthesized the porous Fe₃O₄ flower-like nanostructures successfully, but the optimal RL value was only -28.31 dB and the effective absorption bandwidth (RL < -10 dB) was 3.8 GHz, which also needed to be broadened.

In this study, porous Fe₃O₄/C composite flowers were prepared via solvothermal method followed by a carbon reduction process. These composite flowers exhibit a three-dimensional flower-like shape that was composed of core-shell Fe₃O₄/C nanoparticles coated with amorphous carbon sheets. The microstructures and compositions of the porous Fe₃O₄/C composite flowers were studied by X-ray diffraction (XRD), field emission scanning electron microscopy (FE-SEM), transmission electron microscope (TEM). The carbon content in the porous Fe₃O₄/C composite flowers was evaluated by thermogravimetric analysis. The EM parameters including complex permittivity and complex permeability were measured from 1.0 to 18.0 GHz. The resultant 3D porous Fe₃O₄/C composite flowers present an enhanced EM wave absorption performance comparing with previously reported pure Fe₃O₄ and other Fe₃O₄/C composites and the mechanisms for the enhanced EMI shielding were clarified in terms of synergistic effects between Fe₃O₄ and carbon as well as the structural features.

EXPERIMENTAL SECTION

Materials. Ferric chloride (FeCl₃·6H₂O), urea, ethylene glycol and pyrrole were obtained from Sinopharm Chemical Reagent Co., Ltd. All chemicals were analytical grade and used without further purification.

Preparation of 3D Porous Fe₃O₄/C Composite Flowers. The 3D porous Fe₃O₄/C composite flowers were prepared by adopting a two-step strategy including a facile synthesis of Fe₂O₃ flowers as an intermediate and a following carbon reduction process by using pyrrole as carbon source. In a typical procedure, 0.5 mmol ferric chloride and 3.0 mmol urea were dissolved in 25 mL of ethylene glycol, and the solution was constantly stirred for 1 h at room temperature. Subsequently, the formed homogeneous solution was transferred into a Teflon-lined stainless-steel autoclave (60 mL capacity), which was maintained at 160 °C for 10 h and then allowed to cool down to room temperature. After reaction, the yellow precursor was collected by centrifugal separation, then washed with deionized water for several times and finally dried in a vacuum oven at 60 °C for overnight. The Fe₂O₃ intermediate was obtained by calcinating the as-synthesized precursors in air at 400 °C for 1 h. The final product was obtained through a carbon reduction process by mixing 1 g of as-obtained Fe₂O₃ powders and 0.5 mL of pyrrole in a sealed steel autoclave and heated at 550 °C for 5 h.

Preparation of Samples for Electromagnetic Measurements. In order to investigate the EM properties of the as-synthesized 3D porous Fe₃O₄/C composite flowers, the samples were fabricated by homogeneously dispersing the products into epoxy resin with mass ratio of 50 wt %. Then the Fe₃O₄/C-epoxy resin composites were cut into a toroidal-shaped specimen with an outer diameter of 7.00 mm and an inner diameter of 3.04 mm for EM measurements.

CHARACTERIZATIONS

The compositions and crystal structures of the Fe₃O₄/C composite flowers were characterized by X-ray diffraction (XRD) instrument with Cu K radiation ($\lambda = 0.15406$ nm) at 40 mA and 40 kV ranging from 10° to 80°. The surface morphology of the products and distribution of the elements were observed by a field emission scanning electron microscopy (FE-SEM, JSM-6700F) equipped with an energy dispersive spectrometer (XFlash 5030, Bruker, Germany). The microstructure of the 3D porous Fe₃O₄/C composite flowers was analyzed by a JEOL JEM-2100 transmission electron microscopy (TEM) operating at an accelerating voltage of 200 kV. In order to evaluate the carbon content in Fe₃O₄/C composites, thermogravimetric analysis was carried out on a SDT Q600 analyzer from room temperature to 800 °C at a heating rate of 10 °C min⁻¹. Nitrogen adsorption-desorption isotherms were recorded at 77 K on a QUADRASORB SI-KR/MP (Quantachrome, USA) instrument. The magnetic properties of the products were studied by a vibrating sample magnetometer (VSM, Lake Shore 7400) at room temperature. A network analyzer (Agilent Technologies NS244A) was applied to test the complex permeability (μ_r) and complex permittivity (ϵ_r) of the as-prepared toroidal samples using the coaxial-line method in frequency range of 1–18 GHz.

RESULTS AND DISCUSSION

Scheme 1 presents the synthesis procedures of 3D porous Fe₃O₄/C composite flowers. First, Fe₂O₃ flowers were

Scheme 1. Schematic Illustration of the Synthesis Procedure for 3D Porous Fe₃O₄/C Composite Flowers



obtained via a solvothermal reaction and a following calcination treatment in air. Then, 3D porous Fe₃O₄/C composite flowers were produced after the carbon reduction process by using pyrrole as the carbon source. During this process, carbon was connected together to form carbon sheets and the reduced Fe₃O₄ particles grew at the carbon sheets gradually with increasing the temperature. Hence, pyrrole played two roles in the formation of 3D porous Fe₃O₄/C composite flowers. On one hand, it acted as the reductant. Carbon was generated from the pyrolysis of pyrrole with increasing the temperature to 550 °C and reduced Fe₂O₃ to Fe₃O₄, during which a carbon shell was coated on Fe₃O₄

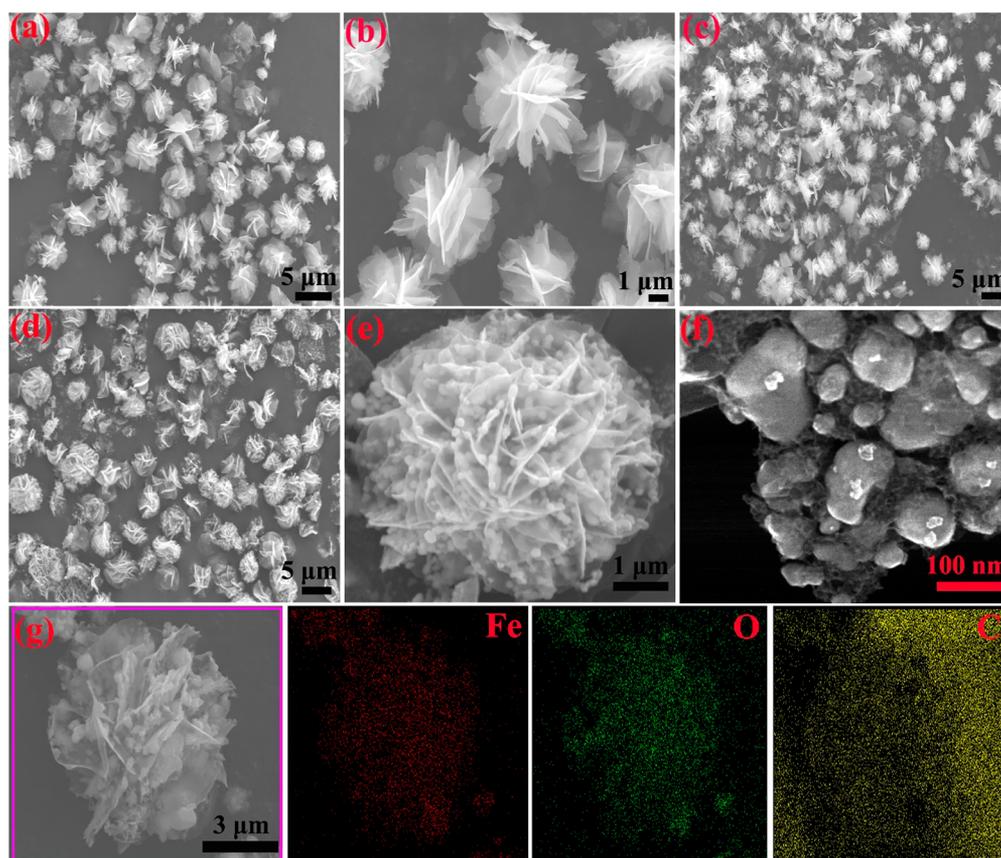


Figure 1. SEM images of the as-synthesized precursor (a and b), Fe_2O_3 flowers (c), 3D porous $\text{Fe}_3\text{O}_4/\text{C}$ composite flowers (d and e), (f) petal surfaces for 3D porous $\text{Fe}_3\text{O}_4/\text{C}$ composite flowers; (g) elemental mappings of 3D porous $\text{Fe}_3\text{O}_4/\text{C}$ composite flowers.

particles. On the other hand, it served as a carbon source for the carbon deposition to form carbon sheets.

Figure 1a,b reveals the morphology of the precursor after the solvothermal reaction. As observed in Figure 1a, the precursor displays a monodisperse 3D flower-like morphology with an average size of 4–6 μm . The precursor is observed to be self-assembled by numerous smooth ultrathin petals as shown in the amplified image, Figure 1b. After calcination treatment in air, no apparent variations can be detected in the shape of Fe_2O_3 flowers (Figure 1c). Figure 1d exhibits the SEM images of 3D porous $\text{Fe}_3\text{O}_4/\text{C}$ composite flowers. No obvious distinctions were observed in the overall size for $\text{Fe}_3\text{O}_4/\text{C}$ flowers but with a great number of particles dispersed on the petal surface, and the dispersed particle sizes display a quite nonuniform distribution from tens of nanometers to hundreds of nanometers since the grain size can be greatly influenced by the heat treatment or solution temperature.^{22,23} An individual $\text{Fe}_3\text{O}_4/\text{C}$ flower was composed of two-dimensional flakes and numerous particles observing from the amplified image in Figure 1e. Taking a closer observation at the petal surfaces of 3D porous $\text{Fe}_3\text{O}_4/\text{C}$ composite flowers in Figure 1f, the particle sizes vary between 20 and 150 nm. In addition, a lot of pores and defects were present on the sheets. To illustrate the spatial distribution of Fe, O and C, the elemental mapping was carried out on an individual $\text{Fe}_3\text{O}_4/\text{C}$ flower, Figure 1g. The obtained results demonstrate that the Fe_3O_4 particles spread all over the structures and display a highly homogeneous dispersion with the presence of carbon.

Figure 2a shows the XRD patterns of the as-synthesized precursor, Fe_2O_3 flowers and 3D porous $\text{Fe}_3\text{O}_4/\text{C}$ composite

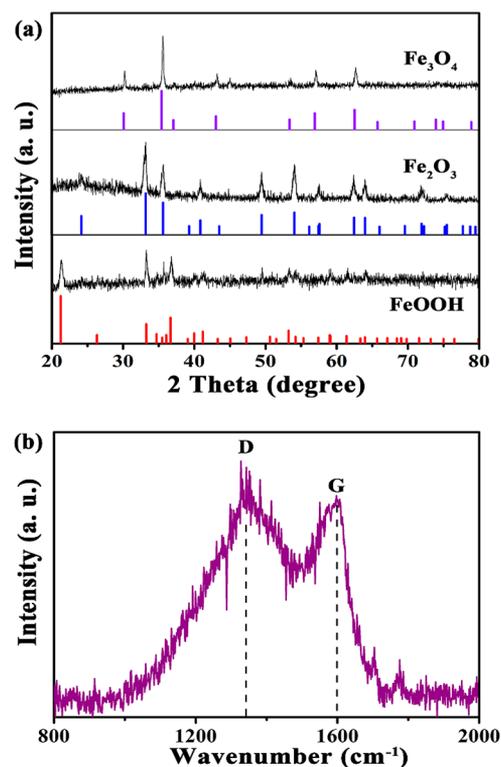


Figure 2. (a) XRD patterns of the as-synthesized precursor, Fe_2O_3 intermediate and 3D porous $\text{Fe}_3\text{O}_4/\text{C}$ composite flowers and (b) Raman spectrum of the $\text{Fe}_3\text{O}_4/\text{C}$ composite flowers.

flowers. The diffraction peaks of the precursor can be assigned to FeOOH (JCPDS No. 29-0713). After the calcination, the diffraction peaks at 24°, 33°, 36°, 49°, 54°, 62° and 64° corresponding to Fe₂O₃ intermediate (JCPDS No. 80-2377) can be clearly seen, indicating the conversion of FeOOH to Fe₂O₃ during the heat treatment. When reduced by carbon, all diffraction peaks at 30°, 35°, 57° and 62° can be typically indexed to Fe₃O₄ (JCPDS Card No. 19-0629), suggesting the complete reduction of Fe₂O₃ to Fe₃O₄. By using the Scherrer equation, the average grain size of the Fe₃O₄ particles is calculated to be 95.6 nm. No other diffraction peaks have been detected, implying the Fe₃O₄/C flowers are almost without oxidation due to the protection of carbon shell. In addition, no diffraction peaks of carbon were observed in the XRD pattern, indicating that the carbon may exist mainly in an amorphous state. The information about the structure of carbon in the Fe₃O₄/C composite flowers is further investigated by Raman analysis (Figure 2b). Two characteristic peaks centered at ca. 1340 and 1595 cm⁻¹ were observed, which respectively correspond to the D band representing amorphous carbon and the G band originating from graphite carbon.⁷ Thus, the intensity ratio between D band and G band (I_D/I_G) can be used to evaluate the graphitization degree. The I_D/I_G value for the Fe₃O₄/C composite flowers is calculated to be 1.16, demonstrating a low graphitization degree, which is in accordance with the XRD results.

The detailed structure is further confirmed by TEM observations (Figure 3a–d). The Fe₃O₄/C composite flowers

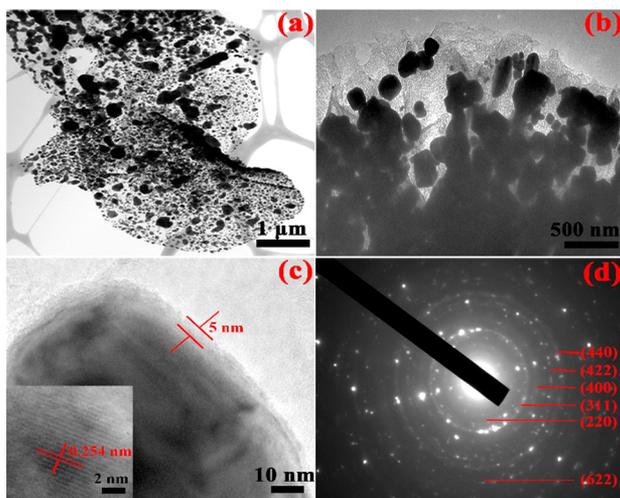


Figure 3. (a, b) TEM, (c) HR-TEM images and (d) SAED patterns of 3D porous Fe₃O₄/C composite flowers. The inset is the lattice spacing taken from the particles in Figure 3c.

are observed to be constructed by two-dimensional sheets and numerous particles with the size in the range of 20–300 nm (Figure 3a). In Figure 3b, the two-dimensional sheets are confirmed to be amorphous carbon sheets, which is in accordance with the XRD results. Obviously, the carbon sheets display a porous characteristic, Figure 3b, attributing to the release of carbon oxides during the high temperature carbon reduction process.²⁴ To further confirm the structure and compositions of Fe₃O₄/C flowers, the HR-TEM image of an individual particle is shown in Figure 3c. On one hand, the interplanar spacing of 0.254 nm can be assigned to the (311) plane of magnetite (the inset Figure 3c), indicating the

formation of Fe₃O₄ core. On the other hand, a carbon layer of ca. 5 nm distributing around Fe₃O₄ particles can be detected, indicating a core–shell structure. The selected-area electron diffraction (SAED) pattern of 3D porous Fe₃O₄/C composite flowers is shown in Figure 3d. The diffraction rings can be well assigned to the (220), (311), (400), (422), (440) and (622) planes of Fe₃O₄,⁸ which is in accordance with XRD results, further confirming the successful synthesis of 3D porous Fe₃O₄/C composite flowers.

To evaluate the weight percentage of carbon in the 3D porous Fe₃O₄/C composite flowers, TGA measurement was conducted in air from room temperature to 800 °C. As shown in Figure 4, the weight loss of ca. 1.1% observed below 150 °C

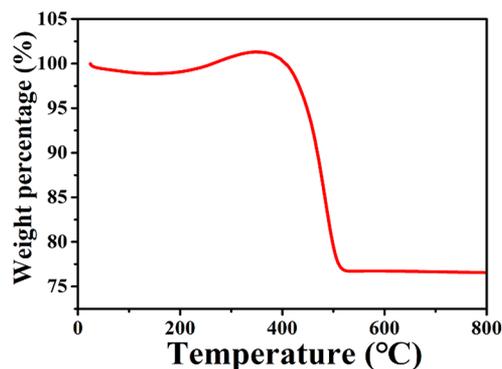


Figure 4. TGA curves of the 3D porous Fe₃O₄/C composite flowers measured in air from room temperature to 800 °C.

is ascribed to the evaporation of water. An intense weight loss of 22.1% occurred in the temperature range of 150–500 °C indicates a simultaneous process of the combustion of carbon and oxidation of Fe₃O₄ to Fe₂O₃. As carbon can be oxidized completely in air at 800 °C, the total 76.8% weight (100%–1.1%–22.1%) should only be the left Fe₂O₃. Based on the above analysis, the content of carbon in the 3D porous Fe₃O₄/C composite flowers can be calculated by Equation 1:

$$\text{wt}\%_{\text{L}} = \frac{3(1 - \text{wt}\%_{\text{carbon}} - \text{wt}\%_{\text{water}})}{2M_{\text{Fe}_3\text{O}_4}} \times M_{\text{Fe}_2\text{O}_3} \quad (1)$$

where wt %_L is the weight percentage of left Fe₂O₃, and M represents molecular weight of the chemicals. The weight ratio of carbon in the porous Fe₃O₄/C composite flowers is calculated to be ca. 24.6 wt %, and Fe₃O₄ content is about 76.4 wt %.

Porous structures involving low density and high specific surface areas are found to be beneficial for improving the EM wave absorption performances and also present in the 3D porous Fe₃O₄/C composite flowers. In order to confirm this porous characteristic, the nitrogen adsorption–desorption isotherms and pore size distributions of the 3D porous Fe₃O₄/C composite flowers are displayed in Figure 5. The measured curves exhibit a similar IV-type isotherm with a long and narrow hysteresis loop at relative pressure from 0.4 to 1.0, implying the presence of mesopores in the 3D porous Fe₃O₄/C composite flowers.^{25,26} Observing from the pore size distribution curve (inset of Figure 5), the pores size of ca. 25 nm takes the dominant position. The BET specific surface area of the 3D porous Fe₃O₄/C composite flowers are 70.2 m² g⁻¹ is larger than that of Fe₃O₄/C nanorings (32.46 m² g⁻¹),⁸

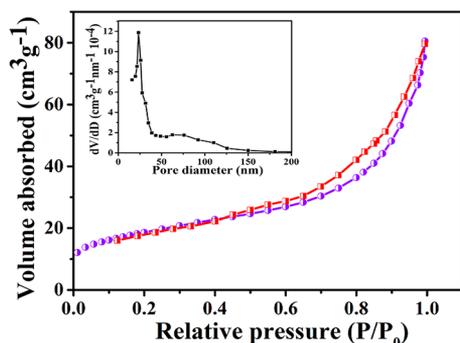


Figure 5. Nitrogen adsorption/desorption isotherms and inset is the corresponding pore size distribution curves of the 3D porous Fe₃O₄/C composite flowers.

porous Fe₃O₄/C nanorods (45.2 m² g⁻¹),¹⁶ mesoporous interconnected C-encapsulated Fe₃O₄ (51.72 m² g⁻¹).²⁷

Figure 6 shows the magnetic hysteresis loops of the 3D porous Fe₃O₄/C composite flowers. The saturation magnet-

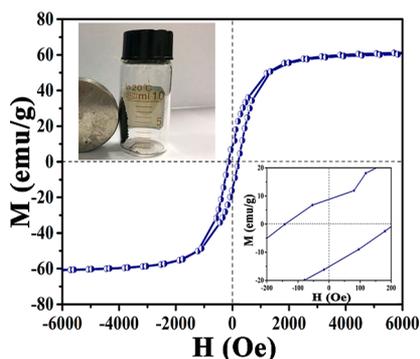


Figure 6. Magnetic hysteresis loop of the 3D porous Fe₃O₄/C composite flowers measured at room temperature. Inset of Figure 6 in the lower right-hand corner shows expanded low field hysteresis curves. Inset in the upper left-hand corner shows the rapid separation of 3D porous Fe₃O₄/C composite flowers by a magnet.

ization (M_s) value for the 3D porous Fe₃O₄/C composite flowers is 61.9 emu g⁻¹. As an indicator of the ferromagnetic property and high magnetization value, a rapid separation by a magnet has been observed in the upper left-hand corner of Figure 6. The M_s is lower than that of bulk Fe₃O₄ (92 emu g⁻¹)¹⁶ and the decreased M_s can be ascribed to the addition of nonmagnetic carbon. The coercive force (H_c) observed from the expanded low field hysteresis curves of inset Figure 6 shows that the 3D porous Fe₃O₄/C composite flowers own a large H_c value of 215.8 Oe. It is widely accepted that the H_c value is strongly related to particle size, shape and magneto-crystalline anisotropy.^{28–30} In this study, the enhanced coercivity value is owing to the larger surface anisotropic energy induced by the large surface of 3D Fe₃O₄/C flowers.^{19,31} Similar enhancement in H_c has been reported in many flower-like, or hierarchical structures.^{18,19,32–34}

To reveal the EM wave absorption properties of the epoxy resin composites with 50 wt % 3D porous Fe₃O₄/C composite flowers, the RL was calculated from the measured complex permittivity and complex permeability at a given absorber thickness and frequency based on the transmit line theory, which can be expressed as follows:^{35,36}

$$Z_{in} = Z_0(\mu_r/\epsilon_r)^{1/2} \tanh\{j(2\pi fd/c)(\mu_r\epsilon_r)^{1/2}\} \quad (2)$$

$$RL = 20\log|(Z_{in} - Z_0)/(Z_{in} + Z_0)| \quad (3)$$

where Z_0 refers to the input impedance of free space, Z_{in} is the input impedance of the absorber, f is the frequency, c is the velocity of light and d is the absorber thickness. In general, a RL value less than -10 dB (RL < -10 dB) is analogous to 90% EM absorption and the corresponding absorption frequency range with RL < -10 dB in one thickness is considered as effective bandwidth. Figure 7a shows the three-dimensional RL curves of the epoxy resin composites containing 50 wt % 3D porous Fe₃O₄/C composite flowers at various thicknesses over frequency range of 1–18 GHz. For an absorber thickness of 4.27 mm, the 3D porous Fe₃O₄/C composite flowers show an optimal RL value of -54.6 dB at 5.7 GHz, and the effective absorption bandwidth is 2.8 GHz (Figure 7b). When the thickness is decreased to 2.1 mm, the minimum RL value of -22.2 dB at 13.6 GHz is observed in Figure 7b and the achieved bandwidth with RL less than -10 dB is 6.0 GHz (12.0–18.0 GHz) as observed in Figure 7c. Besides, it can be found from Figure 7b that with a decrease in the thickness, the absorption peaks will shift to high frequency region. In order to explain this shift, the quarter-wavelength model $t_m = n\lambda/4 = nc/(4f_m\sqrt{|\mu_r\epsilon_r|})$ ($n = 1, 3, 5, \dots$) are employed,³⁷ in which the absorber thickness (t_m) is inversely proportional to the peak frequency (f_m). Figure 7d presents the simulation of the absorber thickness versus peak frequency based on the quarter-wavelength ($\lambda/4$) condition. The red star in Figure 7d denotes the optimal thickness obtained directly from the 2D RL curves, Figure 7b. It is clear that all the red stars are located around the $t_m = \lambda/4$ curve, suggesting the relationship between matching thickness and frequency for the 3D porous Fe₃O₄/C composite flowers can be well-explained by the quarter-wavelength theory.

Compared with previously reported pure Fe₃O₄ absorber, the as-synthesized 3D porous Fe₃O₄/C composite flowers display great advantages, such as stronger absorption intensity, broader absorption bandwidth and thinner thickness. For example, the Fe₃O₄ nanoparticles,³⁸ Fe₃O₄ nanocrystals³⁹ and porous Fe₃O₄ flower-like nanostructures²¹ exhibit a minimum RL value of -21.2, -21.1 and -28.3 dB, respectively, which is much lower than that of 3D porous Fe₃O₄/C composite flowers in this work (-54.6 dB). Such enhanced EM wave absorption performances of 3D porous Fe₃O₄/C composite flowers compared with pure Fe₃O₄ is ascribed to the improved impedance matching level resulted from an effective combination of dielectric loss from carbon and magnetic loss from Fe₃O₄ particles. For other reported Fe₃O₄/C composite absorbers, such as Fe₃O₄/C core-shell composites,¹⁵ RGO-Fe₃O₄ composites,⁴¹ Fe₃O₄/carbon core/shell nanorods,¹⁶ the 3D porous Fe₃O₄/C composite flowers also show stronger absorption capacity and broader absorption bandwidth. The detailed comparisons between the 3D porous Fe₃O₄/C composite flowers in this work and previously reported pure Fe₃O₄ absorber and other Fe₃O₄/C composites are listed in Table 1. Summarizing the comparative results in Table 1, the as-synthesized 3D porous Fe₃O₄/C composite flowers can be considered as an ideal absorption material with strong absorption ability, broad absorption bandwidth and thin absorber thickness.

In order to explore the intrinsic mechanisms for such excellent microwave absorption performances of the 3D

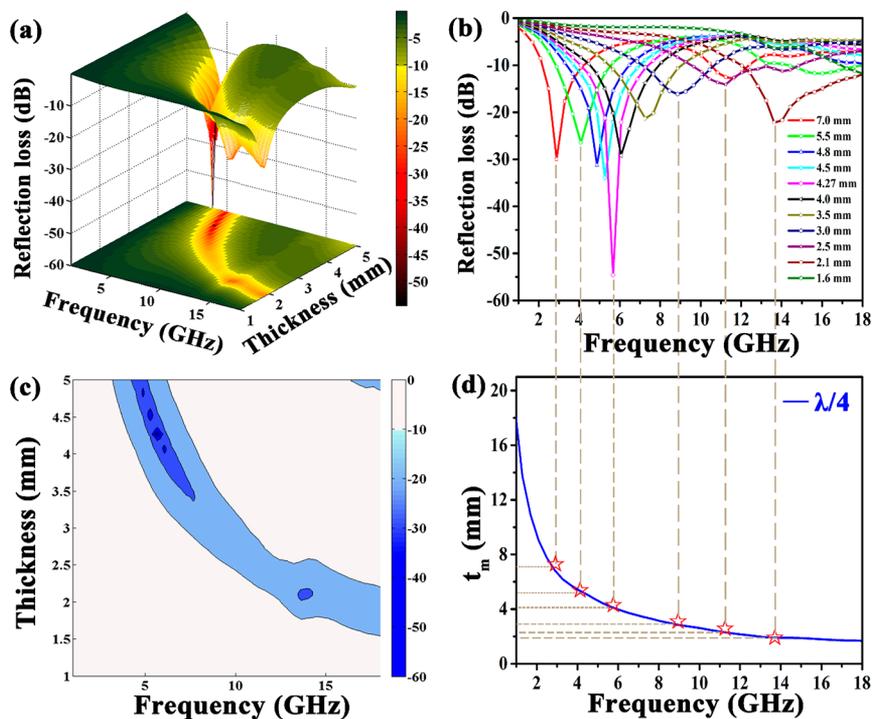


Figure 7. (a) 3D representations of calculated RL values, (b) frequency dependence of RL curves. (c) 2D contour plots of 3D RL representations and (d) simulations of the absorber thickness (t_m) versus peak frequency (f_m) for 3D porous $\text{Fe}_3\text{O}_4/\text{C}$ composite flowers.

Table 1. EM Wave Absorption Properties of Previously Reported Pure Fe_3O_4 Absorber and Other $\text{Fe}_3\text{O}_4/\text{C}$ Composites

Sample	Filling ratio	Optimal RL (dB)	Optimal Thickness (mm)	Bandwidth		ref
				Thickness (mm)	RL < -10 dB (GHz)	
Fe_3O_4 nanoparticles	60 wt %	-21.2	2.0	2.0	3.6 (14.4–18.0)	38
Fe_3O_4 nanocrystals	30 vol%	-21.1	5.0	2.0	–	39
porous Fe_3O_4 flower-like nanostructures	50 wt %	-28.3	2.0	2.0	3.8 (11.7–15.5)	21
$\text{Fe}_3\text{O}_4/\text{C}$ nanospindles	60 wt %	-36.5	5.0	2.0	2.3 (10.5–12.8)	7
core-shell $\text{Fe}_3\text{O}_4/\text{C}$ composites	50 wt %	-36.5	5.0	2.0	3.7 (11.8–15.5)	15
$\text{Fe}_3\text{O}_4/\text{carbon}$ core/shell nanorods	55 wt %	-27.9	2.0	2.0	5.0 (13.0–18.0)	16
$\text{Fe}_3\text{O}_4@\text{C}$ nanotubes	66.7 wt %	-22.6	1.7	2.0	4.8 (11.4–16.2)	40
RGO- Fe_3O_4 composites	40 wt %	-26.4	4.0	2.0	2.8 (10.4–13.2)	41
MWCNT/ Fe_3O_4 hybrid materials	70 wt %	-18.2	2.0	2.0	–	42
3D porous $\text{Fe}_3\text{O}_4/\text{C}$ composite flowers	50 wt %	-54.6	4.27	2.1	6.0 (12.0–18.0)	This work

porous $\text{Fe}_3\text{O}_4/\text{C}$ composite flowers, the electromagnetic parameters including complex permittivity ($\epsilon_r = \epsilon' - j\epsilon''$) and complex permeability ($\mu_r = \mu' - j\mu''$) were investigated, where the real parts (ϵ' and μ') and imaginary parts (ϵ'' and μ'') stand for the storage and loss capability of electric and magnetic energy, respectively. As displayed in Figure 8a, the ϵ' values tend to decrease continuously with negligible fluctuations from 9.51 to 5.32 while ϵ'' decreases gradually from the maximum value of 2.65 at 13.8 GHz for 3D porous $\text{Fe}_3\text{O}_4/\text{C}$ composite flowers with the increasing frequency. Meanwhile, a distinguishable peak was observed in the ϵ'' curves at 13.8 GHz is associated with a dielectric polarization behavior (e.g., ionic polarization, electric polarization, dipolar polarization, interface polarization etc.).^{12,37} As ionic polarization, and electron polarization generally exist in much higher frequency region (10^3 – 10^6 GHz), which could be easily excluded easily.⁴³ To clarify the reasons for the presence of the resonance peak in ϵ'' curves and understand the dielectric loss mechanisms of 3D porous $\text{Fe}_3\text{O}_4/\text{C}$ composite flowers, the Debye dipolar

relaxation model was employed.⁸ According to the Debye dipolar relaxation, the complex permittivity can be described as equation 4:⁴⁴

$$\epsilon_r = \epsilon' - j\epsilon'' = \epsilon_\infty + (\epsilon_s - \epsilon_\infty)/(1 + j\omega\tau) \quad (4)$$

in which ω ($\omega = 2\pi f$), ϵ_s , τ and ϵ_∞ represent the angular frequency, static permittivity, relaxation time and relative permittivity at the high-frequency limit. Based on the above equation 4, ϵ' and ϵ'' could be deduced that

$$\epsilon' = \epsilon_\infty + (\epsilon_s - \epsilon_\infty)/(1 + \omega^2\tau^2) \quad (5)$$

$$\epsilon'' = \omega\tau + (\epsilon_s - \epsilon_\infty)\omega\tau^2/(1 + \omega^2\tau^2) \quad (6)$$

The continuous decrease for the ϵ' curve in Figure 8a with frequency can be well explained by using Equation 5, in which the ϵ' value is inversely proportional to frequency. According to Equations 5 and 6, the relationship between ϵ' and ϵ'' can be expressed as

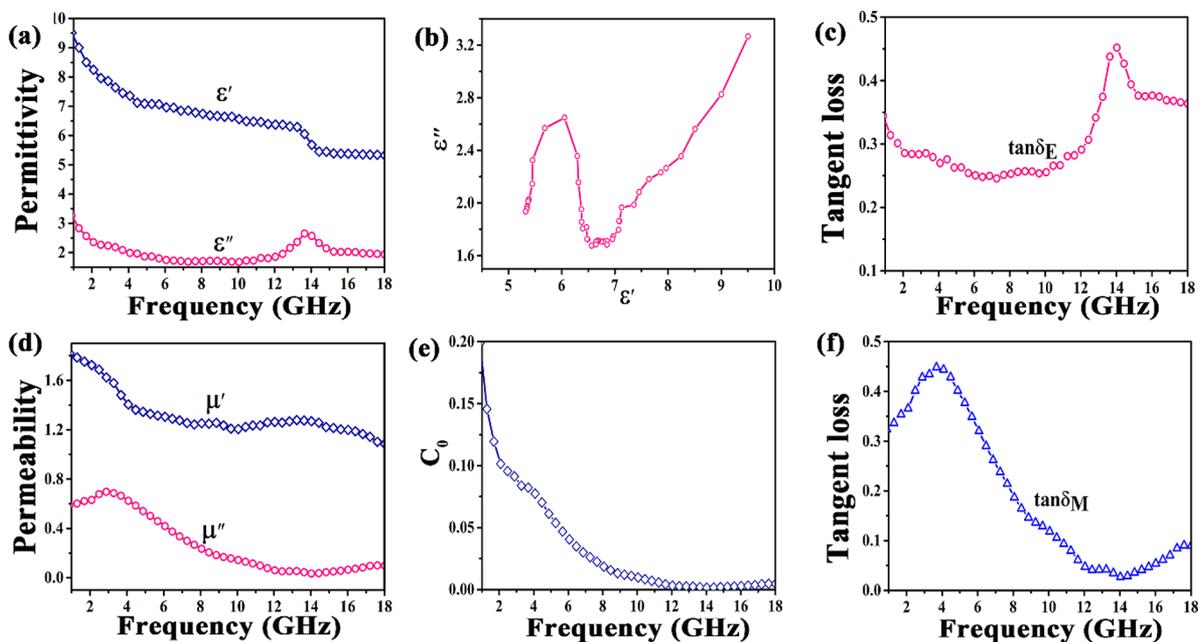


Figure 8. Frequency dependence of (a) complex permittivity, (c) dielectric tangent loss, (d) complex permeability, (e) C_0 values (representing eddy current loss), (f) magnetic tangent loss for 3D porous $\text{Fe}_3\text{O}_4/\text{C}$ composite flowers. (b) Cole–Cole semicircles (ϵ' versus ϵ'').

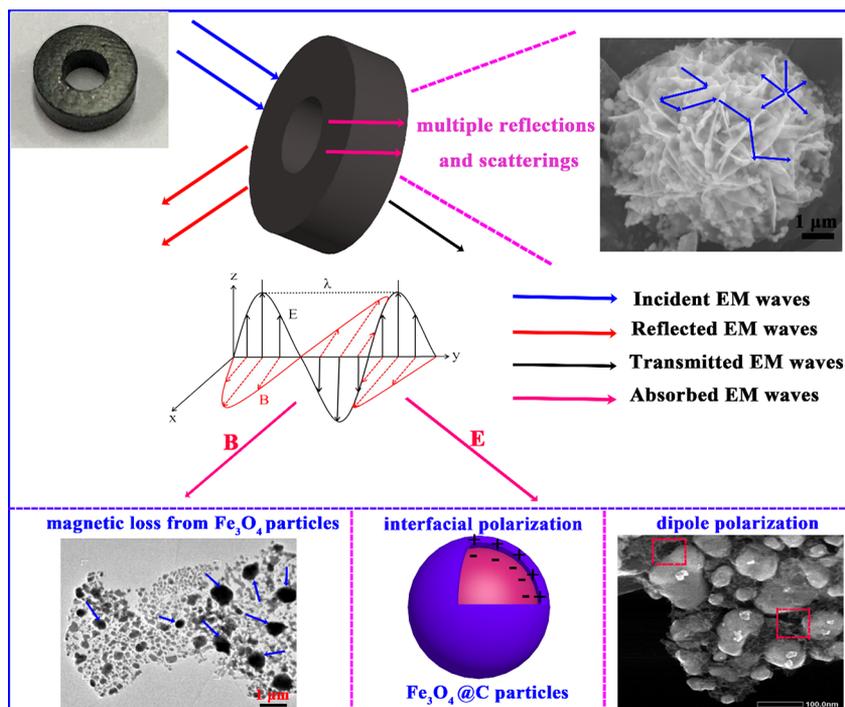


Figure 9. Schematic illustration of the EM wave absorption mechanisms for 3D porous $\text{Fe}_3\text{O}_4/\text{C}$ composite flowers.

$$[\epsilon' - (\epsilon_s + \epsilon_\infty)/2]^2 + (\epsilon'')^2 = [(\epsilon_s - \epsilon_\infty)/2]^2 \quad (7)$$

If ϵ' and ϵ'' of an absorber satisfy Equation 7, the plot of ϵ' against ϵ'' should be a single semicircle, known as the Cole–Cole semicircle, corresponding to one Debye dipolar relaxation.^{3,45} Figure 8b shows the curves of ϵ' plotted against ϵ'' for 3D porous $\text{Fe}_3\text{O}_4/\text{C}$ composite flowers, in which one semicircle was present, demonstrating the existence of Debye dipolar relaxation process. For the as-synthesized 3D porous $\text{Fe}_3\text{O}_4/\text{C}$ composite flowers, the Debye dipolar relaxation process mainly originates from the following aspects. The first

is the interfacial polarization, which exists in the interface between Fe_3O_4 core and carbon shell. Owing to the presence of heterogeneous interfaces, the charges will accumulate at the interfaces and produce an electric dipole moment, which results in interfacial relaxation.^{45,46} Second, the abundant defects existed on the surface of carbon shell and carbon sheet can act as polarization centers, which will generate polarization relaxation under the electromagnetic field.^{47,48} Figure 8d shows the frequency dependence of complex permeability for the epoxy resin composites containing 50 wt % 3D porous $\text{Fe}_3\text{O}_4/\text{C}$ composite flowers. As observed, the μ' values decrease from

1.82 to 1.08 over 1–18 GHz while μ'' exhibits two resonance peaks between 2.0 and 8.0 GHz and 14.0–18.0 GHz, respectively. Generally, the magnetic loss often occurs from the relaxation process during magnetization, which mainly consists of domain wall resonance, exchange resonance, eddy current and natural resonance.⁴⁹ As the domain wall resonance usually occurs in the low frequency range (<2 GHz), which can be excluded first.⁴⁹ The eddy current loss is related to the thickness (d) and electric conductivity (σ) of the materials, which can be expressed by^{49,50}

$$C_0 = \mu''(\mu')^{-2}f^{-1} = 2\pi\mu_0d^2\sigma \quad (8)$$

Supposing that the magnetic loss totally originates from eddy current loss, then the value of C_0 should be constant when frequency varies. It can be seen from Figure 8e that the C_0 varies with increasing frequency, indicating that the eddy current can be suppressed effectively and the eddy current loss is not the main origin of magnetic loss. Concluded from the above analysis, the natural resonance and exchange resonance should be the dominant magnetic loss mechanisms for 3D porous Fe₃O₄/C composite flowers. According to reports, the resonance peaks of μ'' at low frequency from 2.0 to 8.0 GHz should belong to natural resonance while the high frequency resonance peaks from 14.0 to 18.0 GHz can be attributed to exchange resonance.⁴⁹

It is widely known that the dielectric loss and magnetic loss are two key attenuation mechanisms for EM wave absorption. In order to determine which one is dominating for the 3D porous Fe₃O₄/C composite flowers, the dielectric loss tangent and magnetic loss tangent are calculated and displayed in Figure 8c,f. From 1.0 to 8.0 GHz, the $\tan \delta_M$ (0.32–0.46) is close to $\tan \delta_E$ (0.25–0.35), inferring that both magnetic loss and dielectric loss contributes to EM wave absorption in this frequency region. From 8.0 to 18.0 GHz, the $\tan \delta_M$ values decrease rapidly while the $\tan \delta_E$ curve shows an increasing trend, which suggests that dielectric loss is the main mechanism for EM absorption rather than magnetic loss in the high frequency range. The dielectric loss as the dominant mechanism at high frequency range has been reported in various kinds of magnetic carbon composites, such as Fe₃O₄@C core-shell nanotubes,⁴⁰ RGo-Fe₃O₄ composite,⁴¹ Fe/C nanocubes⁵¹ and GO/CNT-Fe₃O₄ composites.⁵²

Based on the above analysis, the enhanced EM wave absorption performances of the 3D porous Fe₃O₄/C composite flowers can be understood by the following aspects shown in Figure 9. First, owing to the synergistic effect of magnetic loss from Fe₃O₄ and dielectric loss from carbon, a good impedance matching can be obtained (shown in Figure 8c,f), endowing the 3D porous Fe₃O₄/C composite flowers with enhanced EM wave absorption properties. The dipolar polarization and interfacial polarization are the main causes of dielectric loss while the magnetic loss mainly originates from natural resonance and exchange resonance. Finally, the 3D porous Fe₃O₄/C composite flowers with large surface area promote the production of multiple reflections, which prolongs the travel path and attenuates the incident wave energy gradually.^{10,53} The porous structure is beneficial to modify the permittivity and improve the impedance matching level. According to the Maxwell–Garnett (MG) theory, which is shown as below:³⁵

$$\epsilon_{\text{eff}}^{\text{MG}} = \epsilon_1 \frac{(\epsilon_2 + 2\epsilon_1) + 2p(\epsilon_2 - \epsilon_1)}{(\epsilon_2 + 2\epsilon_1) - p(\epsilon_2 - \epsilon_1)} \quad (9)$$

where ϵ_1 and ϵ_2 represent the dielectric constant of the host (solid) and guest (air), p is the volume fraction of guest in the porous materials. As the ϵ_1 value is much larger than ϵ_2 ($\epsilon_2 \approx 1$), with the increase of pore volume, p will increase which leads to the decrease of effective permittivity. In this way, the permittivity of the porous Fe₃O₄/C composite flowers can be tuned. Besides, the pores and defects present on the amorphous carbon sheet and carbon shell of 3D porous Fe₃O₄/C composite flowers serve as scattering centers, providing another mechanism for EM wave attenuation.^{54,55} In conclusion, the as-synthesized 3D porous Fe₃O₄/C composite flowers can be considered as an ideal absorber with strong absorption capacity, thin absorber thickness and broad bandwidth. Besides, owing to wide applications of magnetic carbon materials,^{56–59} the 3D porous Fe₃O₄/C composite flowers can be further extended to other fields.

CONCLUSIONS

3D porous Fe₃O₄/C composite flowers were successfully prepared through a facile strategy including a solvothermal approach and a following carbon reduction process. The epoxy resin composites containing 50 wt % 3D porous Fe₃O₄/C composite flowers exhibit an enhanced EM wave absorption comparing with previously reported pure Fe₃O₄ and other Fe₃O₄/C composites. The improved EM wave absorption of the 3D porous Fe₃O₄/C composite flowers are interpreted from the improved impedance matching resulting from an effective combination of strong magnetic loss from Fe₃O₄ and dielectric loss from carbon, as well as the multiple reflections and scatterings arising from the three-dimensional and porous structural advantages. Besides, the core-shell structure significantly improves the antioxidant ability of Fe₃O₄ particles. This work suggests the 3D porous Fe₃O₄/C composite flowers can be regarded as a potential candidate for enhanced EM wave absorption.

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